

FORM PTO-1390 (REV 5-93)		U.S. DEPARTMENT OF COMMERCE PATENT AND TRADEMARK OFFICE		ATTORNEY'S DOCKET NO. 3259.81131	
TRANSMITTAL LETTER TO THE UNITED STATES DESIGNATED/ELECTED OFFICE (DO/EO/US) CONCERNING A FILING UNDER 35 U.S.C. 371				U.S. APPLICATION NO. (PCT/US 371) 09/308770	
INTERNATIONAL APPLICATION NO. PCT/EP97/06596		INTERNATIONAL FILING DATE 26 November 1997 (26.11.97)		PRIORITY DATE CLAIMED 26 November 1996 (26.11.96)	
TITLE OF INVENTION METHOD FOR PRODUCING ORGANICALLY MODIFIED, PERMANENTLY HYDROPHOBIC AEROGELS					
APPLICANT(S) FOR DO/EO/US SCHWERTFEGER, Fritz					
Applicant herewith submits to the United State Designated/Elected Office (DO/EO/US) the following items and other information:					
<ol style="list-style-type: none"> 1. <input checked="" type="checkbox"/> This is a FIRST submission of items concerning a filing under 35 U.S.C. 371. 2. <input type="checkbox"/> This is a SECOND or SUBSEQUENT submission of items concerning a filing under 35 U.S.C. 371. 3. <input checked="" type="checkbox"/> This express request to begin national examination procedures (35 U.S.C. 371(f)) at any time rather than delay examination until the expiration of the applicable time limit set in 35 U.S.C. 371(b) and PCT Articles 22 and 39(1). 4. <input checked="" type="checkbox"/> A proper Demand for International Preliminary Examination was made by the 19th month from the earliest claimed priority date. 5. <input checked="" type="checkbox"/> A copy of the International Application as filed (35 U.S.C. 371(c)(2)) <ol style="list-style-type: none"> a. <input type="checkbox"/> is transmitted herewith (required only if not transmitted by the International Bureau). b. <input checked="" type="checkbox"/> has been transmitted by the International Bureau. c. <input type="checkbox"/> is not required, as the application was filed in the United States Receiving Office (RO/US). 6. <input checked="" type="checkbox"/> A translation of the International Application into English (35 U.S.C. 371(c)(2)). 7. <input type="checkbox"/> Amendments to the claims of the International Application under PCT Article 19 (35 U.S.C. 371(c)(3)) <ol style="list-style-type: none"> a. <input type="checkbox"/> are transmitted herewith (required only if not transmitted by the International Bureau). b. <input type="checkbox"/> have been transmitted by the International Bureau. c. <input type="checkbox"/> have not been made; however, the time limit for making such amendments has NOT expired. d. <input type="checkbox"/> have not been made and will not be made. 8. <input type="checkbox"/> A translation of the amendments to the claims under PCT Article 19 (35 U.S.C. 371(c)(3)). 9. <input type="checkbox"/> An oath or declaration of the inventor(s) (35 U.S.C. 371(c)(4)). 10. <input type="checkbox"/> A translation of the annexes to the International Preliminary Examination Report under PCT Article 36 (35 U.S.C. 371(c)(5)). 					
Items 11-16 below concern other document(s) or information included:					
<ol style="list-style-type: none"> 11. <input type="checkbox"/> An Information Disclosure Statement under 37 C.F.R. 1.97 and 1.98. 12. <input type="checkbox"/> An Assignment document for recording. A separate cover sheet in compliance with 37 C.F.R. 3.28 and 3.31 is included. 13. <input type="checkbox"/> A FIRST preliminary amendment. <input type="checkbox"/> A SECOND or SUBSEQUENT preliminary amendment. 14. <input type="checkbox"/> A substitute specification. 15. <input type="checkbox"/> A change of power of attorney and/or address letter. 16. <input type="checkbox"/> Other items or information: 					

U.S. APPLICATION NO. (if known, See 37 CFR 1.5)		INTERNATIONAL APPLICATION NO PCT/EP97/06596		ATTORNEY'S DOCKET NO. 3259.81131	
17. <input checked="" type="checkbox"/> The following fees are submitted: Basic National Fee (37 CFR 1.492(a)(1)-(5): Search Report has been prepared by the EPO or JPO \$840.00 International preliminary examination fee paid to USPTO (37 CFR 1.482) \$670.00 No International preliminary examination fee paid to USPTO (37 CFR 1.482), but international search fee paid to USPTO (37 CFR 1.445(e)(2)) \$700.00 Neither International preliminary examination fee (37 CFR 1.482) nor international search fee (37 CFR 1.445(a)(2)) paid to USPTO \$970.00 International preliminary examination fee paid to USPTO (37 CFR 1.482) and all claims satisfied provisions of PCT Article 33(1)-(4) \$ 96.00				CALCULATIONS	PTO USE ONLY
ENTER APPROPRIATE BASIC FEE AMOUNT =				\$	840.00
Surcharge of \$130.00 for furnishing the oath or declaration later than 20 or 30 months from the earliest claimed priority date (37 CFR 1.492(e)).				\$	-0-
CLAIMS	NUMBER FILED	NUMBER EXTRA	RATE		
Total Claims	23 - 20 =	3	X \$ 18.00	\$	54.00
Independent Claims	1 - 3 =	0	X \$ 78.00	\$	-0-
Multiple dependent claims (if applicable)			X \$260.00	\$	260.00
TOTAL OF ABOVE CALCULATIONS =				\$	1,154.00
Reduction by 1/2 for filing by small entity, if applicable. Verified Small Entity statement must also be filed (note 37 CFR 1.9, 1.27, 1.28).				\$	
SUBTOTAL =				\$	1,154.00
Processing fee of \$130.00 for furnishing the English translation later than 20 or 30 months from the earliest claimed priority date (37 CFR 1.492(f)).				\$	-0-
TOTAL NATIONAL FEE =				\$	1,154.00
Fee for recording the enclosed assignment (37 CFR 1.21(h)). The assignment must be accompanied by an appropriate cover sheet (37 CFR 3.28, 3.31). \$40.00 per property.				\$	
TOTAL FEES ENCLOSED =				\$	1,154.00
				Amount to be:	
				refunded	\$
				charged	\$
a. <input checked="" type="checkbox"/> A check in the amount of \$ <u>1,154.00</u> to cover the above fees is enclosed. b. <input type="checkbox"/> Please charge my Deposit Account No. 19-0733 in the amount of \$_____ to cover the above fees. A duplicate copy of this sheet is enclosed. c. <input checked="" type="checkbox"/> The Commissioner is hereby authorized to charge any additional fees which may be required, or credit any overpayment to Deposit Account No. 19-0733. A duplicate copy of this sheet is enclosed.					
NOTE: Where an appropriate time limit under 37 CFR 1.494 or 1.495 has not been met, a petition to revive (37 CFR 1.137(a) or (b)) must be filed and granted to restore the application to pending status.					
SEND ALL CORRESPONDENCE TO: Banner & Witcoff, Ltd. Eleventh Floor 1001 G Street, N.W. Washington, D.C. 20001-4597 Telephone: (202) 508-9100 May 25, 1999				<div style="font-family: cursive; font-size: 1.2em; margin-bottom: 5px;"> <i>Shawn A. Kaga reg. no. 32141</i> </div> <div style="display: flex; justify-content: center; align-items: center;"> <div style="text-align: left; margin-right: 10px;"> SIGNATURE Peter McDermott Registration No. 29,411 </div> <div style="font-family: cursive; font-size: 1.2em; margin-left: 10px;"> <i>for Peter McDermott</i> </div> </div>	

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WO 98/23,67

PCT/EP87/06596

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Description

Specification

Process for the preparation of organically modified, permanently hydrophobic aerogels

The invention pertains to a process for the preparation of organically modified, permanently hydrophobic aerogels.

Aerogels, especially those with porosities above 60% and densities below 0.6 g/cm^3 , have an extremely low thermal conductivity and therefore find use as a thermal insulation material as described, for example, in EP-A-0 171 722.

Aerogels in the wider sense, i.e. in the sense of a "gel with air as the dispersing agent," are prepared by drying a suitable gel. Aerogels in the narrower sense, xerogels and cryogels, are included in the concept of an "aerogel" in this sense. In this connection, a dried gel is termed an aerogel in the narrower sense if the liquid of the gel is removed at temperatures above the critical temperature and starting out from pressures above the critical pressure. If, by contrast, the liquid of the gel is removed subcritically, e.g. with the formation of a liquid/vapor boundary phase, then one designates the produced gel a xerogel.

When using the term aerogels in the present application, one is dealing with aerogels in the wider sense, i.e. in the sense of a "gel with air as the dispersing agent."

In addition, one can basically subdivide aerogels into inorganic and organic aerogels..

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Inorganic aerogels have been known since 1931 (S.S. Kistler, Nature 1931, 127, 741). Since then, aerogels have been produced from the most varied initial materials. For example, SiO_2 , Al_2O_3 , TiO_2 , ZrO_2 , SnO_2 , Li_2O , CeO_2 and V_2O_5 aerogels as well as mixtures of these could be prepared (H. D. Gesser, P. C. Goswami, Chem. Rev. 1989, 89, 765 ff).

For several years, organic aerogels have also been known. One finds in the literature, e.g., organic aerogels based on resorcinol/formaldehyde, melamine/formaldehyde or resorcinol/furfurol (R. W. Pekala, J. Mater. Sci. 1989, 24, 3221, US-A 5,508,341, RD 388,047 [368,047?], WO94/22,943 and US-A-5,556,892). In addition, organic aerogels of polyisocyanates (WO95/03,358) and polyurethanes (US-5,484,818) are also known. One proceeds from initial materials such as formaldehyde and resorcinol dissolved in water, as described, for example, in US-A-5,508,341; these are brought to reaction with one another by suitable catalysts, the water in the pores of the gel that forms is exchanged for a suitable organic solvent, and then the gel is dried supercritically.

Inorganic aerogels can be prepared in different ways.

First of all, SiO_2 aerogels can be produced by acid hydrolysis and condensation of tetraethylorthosilicate in ethanol. A gel is thus formed, which can be dried by supercritical drying while retaining its structure. Production processes based on this drying technique are known, e.g., from EP-A-0 396,076; WO 92/03,378; and WO 95/06,617.

An alternative to the above drying is offered by a process for subcritical drying of SiO_2 gels, in which these are reacted with a silylation agent containing

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chlorine prior to drying. The SiO_2 gel can be obtained, for example, by acid hydrolysis of tetraalkoxysilanes, preferably tetraethoxysilanes (TEOS), in a suitable organic solvent, preferably ethanol, by means of water. After exchange of the solvent for a suitable organic solvent, the obtained gel is reacted with a chlorine-containing silylation agent in an additional step. Methylchlorosilanes ($\text{Me}_{4-n}\text{SiCl}_n$ with $n = 1$ to 3) are thus preferably used as silylation agents, due to their reactivity. The SiO_2 gel that is thus formed and is modified with methylsilyl groups on its surface can then be dried in air from an organic solvent. Thus aerogels with densities below 0.4 g/cm^3 and porosities greater than 60% can be obtained. The production process based on this drying technique is described in detail in WO 94/25,149.

The above-described gels can be reacted with tetraalkoxysilanes in aqueous alcohol solution prior to drying and can be aged in order to increase the gel network strength, as disclosed, e.g., in WO 92/20,623.

The tetraalkoxysilanes used in the above-described process as initial materials, however, represent an extraordinarily high cost factor.

A considerable reduction in cost can be achieved by the use of water glass as the initial material for the production of SiO_2 gels. For this purpose, for example, a silicic acid can be produced from an aqueous water-glass solution by means of an ion-exchange resin, and this acid can be polycondensed by addition of a base to form a SiO_2 gel. After exchange of the aqueous medium for a suitable organic solvent, the obtained gel is then converted with a chlorine-containing silylation agent in an additional step. Methylchlorosilanes ($\text{Me}_{4-n}\text{SiCl}_n$

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with $n = 1$ to 3) are also preferably utilized as silylation agents, due to their reactivity. The modified SiO_2 gel that is formed and is modified with methylsilyl groups on the surface can then also be dried in air from an organic solvent. The production process based on this technique is described in detail, e.g., in DE-A-4,342,548.

Hydrogen chloride (HCl) as well as a multiple number of byproducts combined therewith are necessarily formed in very large quantities in the silylation by means of chlorine-containing silylation agents, and these require sometimes a very expensive and cost-intensive cleaning of the silylated SiO_2 gels by multiple washings with a suitable organic solvent.

The use of a silylation agent free of chlorine is described in DE-C 195 02,453. For this purpose, for example, a silicate-type lyogel produced according to the above-described method is proposed and is reacted with a chlorine-free silylation agent. Preferably methylisopropene oxysilanes ($\text{Me}_{4-n}\text{Si}(\text{OC}(\text{CH}_3)\text{CH}_2)_n$ with $n = 1$ to 3) are used as silylation agents. The thus-formed SiO_2 gel that is modified with methylsilyl groups on the surface can then also be dried in air from an organic solvent.

By the use of chlorine-free silylation agents, in fact, the problem of formation of HCl is solved, but the chlorine-free silylation agents that are used represent a very high cost factor.

In WO 95/06,617 and in the German Patent Application 195-41 279.6, methods are disclosed for the production of silicic acid aerogels with hydrophobic surface groups.

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In WO 95/06,617, silicic acid aerogels are obtained by reaction of a water-glass solution with an acid at a pH value of 7.5 to 11, extensive release of the silicic acid hydrogel that forms from the ionic components by washing with water or dilute aqueous solutions of inorganic bases, whereby the pH of the hydrogel is maintained in the range of 7.5 to 11, expelling the aqueous phase obtained in the hydrogel by an alcohol and subsequent supercritical drying of the obtained alkogel.

In German Patent Application 195-41 279.6, in a way similar to that described in WO 95/06,617, silicic-acid aerogels are produced and are then dried subcritically.

In both methods, however, the omission of chlorine-containing silylation agents leads only to an aerogel with hydrophobic surface groups bound via oxygen. These can easily be cleaved again in a water-containing atmosphere. Thus the described aerogel is only hydrophobic for a short time.

It is also possible to utilize organically modified gels without final drying to the aerogel in the most varied fields of technology, such as, e.g., in chromatography, in cosmetics, and in the pharmaceutical field.

The task of the present invention was thus to prepare a method for the production of permanently hydrophobic aerogels, in which a commercially available, inexpensive silylation agent can be used, without incurring the other disadvantages described above, which are known from the prior art.

This task is resolved by a process for the production of organically modified aerogels with permanently hydrophobic surface groups, in which one

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- a) provides a lyogel,
- b) washes the lyogel provided in step a) with an organic solvent,
- c) surface-silylates the gel obtained in step b), and
- d) dries the surface-silylated gel obtained in step c),

characterized in that in step c), as the silylation agent, a disiloxane of formula I is used



whereby the residues R, independently of one another, the same or different, indicate in each case a hydrogen atom or a nonreactive organic, linear, branched, cyclic, saturated or unsaturated, aromatic or heteroaromatic residue, preferably C₁-C₁₈ alkyl or C₆-C₁₄ aryl, and particularly preferred C₁-C₆ alkyl, cyclohexyl or phenyl, particularly methyl or ethyl.

In the present invention, a lyogel is understood to mean a gel dispersed in at least one solvent. The solvent may also be water. If the water component in the solvent amounts to at least 50%, then one also speaks of a hydrogel.

The network of the lyogel may be present in any organic and/or inorganic base composition. All of the systems of the prior art known to the person skilled in the art can be used as the organic base composition. An inorganic base composition is preferably based on oxidic silicon, tin, aluminum, gallium, indium, titanium and/or zirconium compounds, and particularly preferred are those based on oxidic silicon, aluminum, titanium and/or zirconium compounds. Most preferred is a silicate-type hydrogel, which may contain fractions of zirconium, aluminum, titanium, vanadium and/or iron compounds, particularly a pure silicate-

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type hydrogel. In the case of organic and/or inorganic base compositions, the different components must not necessarily be distributed homogeneously and/or form a continuous network. It is also possible that individual components are present partially or completely in the form of inclusions, individual nuclei and/or agglomerations in the network.

The disiloxanes used according to the invention, when compared with the chlorine-containing silylation agents known from the prior art, have the advantage that no chlorine-containing byproducts are formed. In addition, they can easily be separated from aqueous phases based on their insolubility, which makes possible the recovery of excess reagents. In this way, it is possible to minimize silylation times by the use of excess concentrations.

The preparation of the lyogels provided in step a) can be produced according to all methods known to the person skilled in the art.

Three preferred forms of embodiment for the preparation of silicate-type lyogels will be described in more detail in the following, but without, however limitation to these.

In a first preferred form of embodiment, in step a) a silicate-type lyogel is provided, which is obtained by hydrolysis and condensation of Si alkoxides in an organic solvent with water. A tetraalkoxysilane, preferably tetraethoxy- or tetramethoxysilane is used as the Si alkoxide. The organic solvent is thus preferably an alcohol, and particularly preferred ethanol or methanol, to which up to 20 vol.% water can be added. In the hydrolysis and concentration of Si

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alkoxides in an organic solvent with water, acids and/or bases may be added in a one- or two-step process as catalysts.

The lyogel provided in step a) may also contain zirconium, aluminum, tin and/or titanium compounds suitable for condensation.

In addition, before and/or during the gel preparation, opacifiers can be added as additives, particularly IR opacifiers for reduction of the radiation contribution to the heat conductivity, such as, e.g., carbon black, titanium oxides, iron oxides, and/or zirconium oxides.

In addition, fibers can be added to the sol in order to increase the mechanical stability. Inorganic fibers, such as, e.g., glass fibers or mineral fibers, organic fibers, such as, e.g., polyester fibers, aramide fibers, nylon fibers or fibers of plant origin, as well as mixtures of these can be used as the fiber materials. The fibers may also be coated, such as, e.g., polyester fibers, which are metallized with a metal, such as, e.g., aluminum.

The production of the lyogel is generally conducted at a temperature between the freezing point of the solution and 70°C. In this way, if necessary, a shaping step can be conducted simultaneously, such as, e.g., spray forming, extrusion or drop formation.

The obtained lyogel can also be subjected to an aging. This is generally done between 20°C and the boiling point of the organic solvent. If necessary, aging can also be conducted under pressure at higher temperatures. The time generally amounts to up to 48 hours, preferably up to 24 hours.

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In a second preferred form of embodiment, a silicate-type hydrogel is introduced in step a), which is prepared by bringing an aqueous water-glass solution to a pH of ≤ 3 by means of an acidic ion-exchanger resin, a mineral acid, or a hydrochloric acid solution, then polycondensing the silicic acid that forms thereby by addition of a base to a SiO_2 gel, and, if a mineral acid or a hydrochloric acid solution has been used, the gel is washed with water essentially free of electrolyte. The polycondensation to the SiO_2 gel can be undertaken both in one step as well as in multiple steps.

Preferably sodium and/or potassium water glass are used as the water glass. Preferably an acidic resin is used as the ion-exchanger resin, whereby in particular, those resins are suitable, which contain sulfonic acid groups. If one uses mineral acids, hydrochloric acid and/or sulfuric acid are particularly suitable. If one uses hydrochloric acid solutions, particularly aluminum salts are suitable, especially aluminum sulfate and/or chloride. As the base, generally NH_4OH , NaOH , KOH , $\text{Al}(\text{OH})_3$ and/or colloidal silicic acid are used.

The hydrogel preferably prepared from the above-described silicate-type initial compounds may also contain zirconium, aluminum, tin and/or titanium compounds capable of condensation.

In addition, prior to and/or during the gel production, opacifiers may be added as additives, particularly IR opacifiers, for the reduction of the radiation contribution to the heat conductivity, such as, e.g., carbon black, titanium oxides, iron oxides and/or zirconium oxides.

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In addition, fibers can be added to the sol for increasing the mechanical stability. Inorganic fibers such as, e.g., glass fibers or mineral fibers, organic fibers such as e.g., polyester fibers, aramide fibers, nylon fibers or fibers of plant origin, as well as mixtures of the same can be used as fiber materials. The fibers may also be coated, such as, e.g., polyester fibers, which are metallized with a metal, such as, e.g., aluminum.

The production of the hydrogel is generally conducted at a temperature between the freezing point and the boiling point of the solution. Thus, if necessary, a shaping step can be conducted simultaneously, such as, e.g., spray forming, extrusion or drop formation.

The obtained hydrogel can also be subjected to an aging. This aging may be produced prior to and/or after an above-described possible washing with water, with which the gel is essentially washed free of electrolyte.

The aging is conducted generally at a temperature in the range of 20 to 100°C, preferably at 40 to 100°C and particularly at 80 to 100°C, and at a pH value of 4 to 11, preferably 5 to 9, and particularly 5 to 8. The time for this generally amounts to up to 48 hours, preferably up to 24 hours, and particularly preferably up to 3 hours.

In a third preferred form of embodiment, in step a) a silicate-type hydrogel is provided, which is prepared by obtaining a SiO_2 gel from an aqueous water-glass solution by means of at least one organic and/or inorganic acid through the intermediate step of a silicic acid sol.

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In general, a 6 to 25 wt.% (with respect to the SiO_2 content) sodium and/or potassium water-glass solution is used as the water-glass solution. Preferred is a 10 to 25 wt.% water-glass solution, and particularly preferred is a 10 to 18 wt.% water-glass solution.

In addition, the water-glass solution may also contain up to 90 wt.%, with respect to the SiO_2 , of zirconium, aluminum, tin and/or titanium compounds capable of condensation.

As acids, generally 1 to 50 wt.% acids are used, preferably 1 to 10 wt.% acids. Preferred acids are sulfuric acid, phosphoric acid, hydrofluoric acid, oxalic acid, and/or hydrochloric acid. Particularly preferred is hydrochloric acid. However, mixtures of the corresponding acids may also be utilized.

In addition to the mixing, properly speaking, of the water-glass solution and the acid, it is also possible to introduce a part of the acid into the water-glass solution and/or a part of the water-glass solution into the acid prior to the mixing itself. It is possible in this way to vary the ratio of the material flows of water-glass solution/acid over a very broad range.

After mixing the two solutions, preferably a 5 to 12 wt.% SiO_2 gel is obtained. A 6 to 9 wt.% SiO_2 gel is particularly preferred.

In order to assure an optimal intermixing of the water-glass solution and the acid, before a SiO_2 gel is formed, both solutions should preferably have, independent of one another, a temperature between 0 and 30°C, particularly preferred between 5 and 25°C, and particularly between 10 and 20°C.

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The rapid intermixing of the two solutions is conducted in devices known to the person skilled in the art, such as, e.g., boilers with stirring apparatus, mixing nozzles and static mixers. Preferred are semicontinuous or continuous processes, such as, e.g., mixing nozzles.

If necessary, a forming step can be conducted simultaneously with production, e.g., by spray-forming, extrusion or drop formation.

The obtained hydrogel may also be subjected to an aging. This is generally done at 20 to 100°C, preferably at 40 to 100°C, particularly at 80 to 100°C and a pH value of 2.5 to 11, preferably 5 to 8. The time for this generally amounts to up to 12 hours, preferably up to 2 hours, and particularly preferred, up to 30 minutes.

The prepared gel is preferably washed with water, particularly preferably until the wash water used is free of electrolyte. If an aging of the gel is to be conducted, the washing can be conducted prior to, during and/or after the aging, whereby the gel in this case is preferably washed during or after the aging. For the washing, a part of the water can be replaced by organic solvent. The water content, however, should preferably be high enough that the salts in the pores of the hydrogel do not crystallize out.

In order to remove sodium and/or potassium ions as extensively as possible, the hydrogel can also be washed with a mineral salt prior to, during and/or after the washing with water. Preferred mineral salts are thus also the mineral salts named as preferred for the production of the hydrogel.

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Further, opacifiers can be added to the water glass, the acid and/or the sol as additives, particularly IR opacifiers, for the reduction of the radiation contribution to the heat conductivity such as, e.g., carbon black, titanium oxides, iron oxides and/or zirconium oxides.

In addition, fibers can be added to the water glass, to the acid and/or to the sol in order to increase the mechanical stability. Inorganic fibers, such as, e.g., glass fibers or mineral fibers, organic fibers, such as, e.g., polyester fibers, aramide fibers, nylon fibers or fibers of plant origins as well as mixtures of the same can be used as fiber materials. The fibers may also be coated, such as, e.g., polyester fibers, which are metallized with a metal, such as, e.g., aluminum.

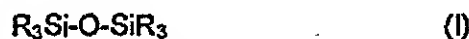
In step b), one washes the gel obtained from step a) with an organic solvent, preferably until the water content of the gel is ≤ 5 wt.%, particularly preferred ≤ 2 wt.% and in particular ≤ 1 wt.%. Generally aliphatic alcohols, ethers, esters or ketones as well as aliphatic or aromatic hydrocarbons are used as solvents. Preferred solvents are methanol, ethanol, acetone, tetrahydrofuran, acetic acid ethyl ester, dioxane, pentane, n-hexane, n-heptane and toluene. Particularly preferred as the solvent is acetone, tetrahydrofuran, pentane and n-heptane. Mixtures of the named solvents may also be used. Further, the water can also be washed out first with a water-miscible solvent, e.g., an alcohol, acetone or THF, and then the latter is washed out with a hydrocarbon. Preferably pentane or n-heptane is used as the hydrocarbon.

The lyogel obtained in step b) may be subjected to an aging. This is generally done between 20°C and the boiling point of the organic solvent. If

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necessary, aging may be conducted also under pressure at higher temperatures. The time generally amounts to up to 48 hours, preferably up to 24 hours. After such an aging, if necessary, another solvent exchange for the same or a different solvent can be conducted. This additional aging step may also be repeated several times.

In step c), the solvent-containing gel is reacted with a disiloxane of formula I as the silylation agent.



whereby the residues R, independently of one another, either the same or different, each time represent a hydrogen atom or a nonreactive organic, linear, branched, cyclic, saturated or unsaturated, aromatic or heteroaromatic residue, preferably C₁-C₁₈ alkyl or C₆-C₁₄ aryl, particularly preferred C₁-C₆ alkyl, cyclohexyl or phenyl, particularly methyl or ethyl.

The solvent-containing gel in step c) is preferably reacted with a symmetric disiloxane, whereby a symmetric disiloxane is understood to be a disiloxane in which both Si atoms have the same residue R.

Particularly preferred, disiloxanes are used in which all residues R are the same. In particular, one uses hexamethyldisiloxane.

The reaction is generally conducted at 20°C up to the boiling point of the silylation agent, if necessary, in a solvent. Preferred solvents here are the solvents described as preferred in step b). Particularly preferred is acetone, tetrahydrofuran, pentane and n-heptane. If the silylation is produced in a solvent,

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then the silylation is generally conducted between 20°C and the boiling point of the solvent.

In a preferred form of embodiment, silylation is conducted in the presence of a catalyst, for example an acid or base. Preferably, acids are utilized as the catalyst. Particularly preferred acids are hydrochloric acid, sulfuric acid, acetic acid and/or phosphoric acid.

In another form of embodiment, the silylation is conducted in the presence of catalytic quantities of a silylation agent, which forms acids in the presence of water. Preferably, chlorosilanes are [used], and particularly preferred is trimethylchlorosilane (TMCS). In addition, a combination of acids or bases and TMCS is also possible.

Prior to step d), the silylated gel is preferably washed with a protic or aromatic solvent, until unreacted silylation agent is essentially removed (residual content ≤ 1 wt.%). Suitable solvents are those named in step b). Analogously, the solvents named there as preferred are also preferred here.

In step d), the silylated and, if necessary, washed gel is preferably dried subcritically, preferably at temperatures from -30°C to 200°C, and particularly preferred, 0 to 100°C, as well as pressures preferably from 0.001 to 20 bars, and particularly preferably 0.01 to 5 bars, particularly 0.1 to 2 bars, for example by radiation, convection and/or contact drying. Drying is preferably conducted until the gel has a residual solvent content of less than 1 wt.%. The aerogels obtained in the drying are permanently hydrophobic.

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The gel obtained in step c) may also be dried supercritically. This requires temperatures higher than 200°C and/or pressures higher than 20 bars, depending on the respective solvent. This is possible without anything further, but it is associated with increased expenditure and does not offer essential advantages.

In another form of embodiment, the gel can be subjected to another network reinforcement, each time depending on application, prior to the silylation in step c). This is done by reacting the obtained gel with a solution of an orthosilicate of formula $R^1_{4-n}Si(OR^2)_n$ capable of condensation, preferably an alkyl and/or aryl orthosilicate, whereby $n = 2$ to 4 and R^1 and R^2 , independently of one another, are hydrogen atoms, linear or branched C_1 - C_8 alkyl, cyclohexyl or phenyl residues, or with an aqueous silicic-acid solution.

In another form of embodiment, after the shaping polycondensation and/or each subsequent process step, the gel can be comminuted according to techniques known to the person skilled in the art, such as, e.g., milling.

The aerogels produced according to the process of the invention find particular use as heat insulation materials.

The process according to the invention is described in more detail in the following, based on examples of embodiment, without thereby being limited to these.

Example 1

2 liters of a sodium water-glass solution (SiO_2 content of 6 wt. % and $Na_2O:SiO_2$ ratio of 1:3.3) are passed through a sheathed glass column (length =

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100 cm, diameter = 8 cm), which is packed with 4 liters of an acidic ion-exchanger resin (styrene-divinylbenzene copolymer with sulfonic acid groups, commercially available under the name ®Duolite C20) (at approximately 70 ml/min). The column is operated at a temperature of approximately 7°C. The silicic-acid solution exiting at the lower end of the column has a pH value of 2.3. This solution is brought to a pH of 4.7 for the polycondensation with a 1.0 molar NaOH solution. After this, the gel that forms is aged for another 3 hours at 85°C and then the water is exchanged for acetone with 3 liters of acetone. Then the acetone-containing gel is silylated with hexamethyldisiloxane at room temperature for 5 hours (2.5 wt.% hexamethyldisiloxane per gram of wet gel). After washing the gel with 3 liters of acetone, drying of the gel is conducted in air (3 hours at 40°C, then 2 hours at 50°C and 12 hours at 150°C). The thus-obtained transparent aerogel has a density of 0.15 g/cm³, a heat conductivity of 15 [16?] mW/mK, a specific surface according to BET of 600 m²/g and is permanently hydrophobic.

Example 2

424 g of a 7.5% HCl solution cooled to 10°C is reacted dropwise with 712 g of a sodium water-glass solution cooled to 10°C (with a content of 13 wt.% SiO₂ and a Na₂O:SiO₂ ratio of 1:3.3). A pH value of 4.7 is thereby adjusted. The hydrogel formed after several seconds is aged for one hour at 85°C. It is then washed with 3 liters of hot water and the water is exchanged for acetone with 3 liters of acetone. Then the acetone-containing gel is silylated with hexamethyldisiloxane (2.5 wt.% hexamethyldisiloxane per gram of wet gel) for 5

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hours at room temperature. After washing the gel with 3 liters of acetone, it is dried in air (3 hours at 40°C, then 2 hours at 50°C and 12 hours at 150°C).

The thus-obtained aerogel has a density of 0.15 g/cm³, a heat conductivity of 17 mW/mK, a specific surface according to BET of 580 m²/g and is permanently hydrophobic.

Example 3

The hydrogel is produced as described in Example 2. The hydrogel aged for one hour at 85°C is then washed with 3 liters of warm water and the water is exchanged for acetone with 3 liters of acetone. Then the acetone-containing gel is silylated with hexamethyldisiloxane (2.5 wt.% hexamethyldisiloxane per gram of wet gel) in the presence of 0.1 wt.% trimethylchlorosilane (0.7 wt.% trimethylchlorosilane per gram of wet gel) for 5 hours at room temperature. After washing the gel with 3 liters of acetone, it is dried in air (3 hours at 40°C, then 2 hours at 50°C and 12 hours at 150°C).

The thus-obtained aerogel has a density of 0.14 g/cm³, a heat conductivity of 16 mW/mK, a specific surface according to BET of 590 m²/g and is permanently hydrophobic.

Example 4

The hydrogel is produced as described in Example 2. The hydrogel aged for 1 hour at 85°C is then washed with 3 liters of warm water and the water is exchanged for acetone with 3 liters of acetone. Then the acetone-containing gel is silylated with hexamethyldisiloxane (2.5 wt.% hexamethyldisiloxane per gram

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of wet gel) in the presence of 0.1 wt.% 1 N aqueous hydrochloric acid (0.1 wt.% 1 N aqueous hydrochloric acid per gram of wet gel) for 5 hours at room temperature. After washing the gel with 3 liters of acetone, it is dried in air (3 hours at 40°C, then 2 hours at 50°C and 12 hours at 150°C).

The thus-obtained aerogel has a density of 0.14 g/cm³, a heat conductivity of 16 mW/mK, a specific BET surface of 570 m²/g and is permanently hydrophobic.

The heat conductivities were measured with a resistance-wire method (see, e.g., O. Nielssen, G. Rüschenpöhler, J. Gross, J. Fricke, High Temperatures – High Pressures, Vol. 21, 267-274 (1989)).

[Patent Claims follow – previously translated]

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Claims

1. Process for the preparation of organically modified aerogels with permanently hydrophobic surface groups in which one

- introduces a lyogel into the reactor;
- washes the lyogel introduced into the reactor in step a) with an organic solvent;
- surface-silylates the gel obtained in step b) and
- dries the surface-silylated gel obtained in step c),

characterized by the feature that, as the silylating agent in step c), one uses a disiloxane of formula I



whereby the residues R, independently of one another, identically or differently, signify in each case a hydrogen atom or a nonreactive, organic, linear, branched, cyclic, saturated or unsaturated, aromatic or heteroaromatic residue.

2. Process in accordance with Claim 1, characterized by the feature that, in step a), one introduces a silicate-type lyogel into the reactor.

3. Process in accordance with Claim 2, characterized by the feature that, in step a), one introduces into the reactor a silicate-type lyogel which is obtainable by hydrolysis and condensation of Si alkoxides in an organic solvent with water.

4. Process in accordance with Claim 2, characterized by the feature that, in step a), one introduces into the reactor a silicate-type hydrogel that is prepared by bringing an aqueous water glass solution to a pH value ≤ 3 with the aid of an acidic ion-exchanged resin or an inorganic acid and, via the addition of a

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base, polycondensing the silicic acid, which is produced in this way, to give a SiO_2 gel and, if an inorganic acid has been used, washing the gel essentially free from electrolytes with water.

5. Process in accordance with Claim 2, characterized by the feature that one introduces into the reactor in step a) a silicate-type gel (which is prepared by obtaining it from an aqueous water glass solution with the aid of at least one organic and/or inorganic acid via the intermediate stage of a silicic acid sol.

6. Process in accordance with one of Claims 1 through 5, characterized by the feature that, prior to and/or during the preparation of the gel, one adds IR turbidity-promoting agents.

7. Process in accordance with one of Claims 1 through 6, characterized by the feature that fibers are added prior to, and/or during, the preparation of the gel.

8. Process in accordance with at least one of the preceding claims, characterized by the feature that one allows the lyogel obtained in step a) to age before it is washed in step b).

9. Process in accordance with at least one of the preceding claims, characterized by the feature that one washes the gel in step b) for a sufficiently long time until the water content of the gel is ≤ 5 wt%.

10. Process in accordance with at least one of the preceding claims, characterized by the feature that use is made of aliphatic alcohols, ethers, esters, or ketones and aliphatic or aromatic hydrocarbons as the organic solvents in step b).

11. Process in accordance with at least one of the preceding claims, characterized by the feature that use is made of a symmetrical disiloxane as the silylating agent in step c).

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12. Process in accordance with at least one of the preceding claims, characterized by the feature that, as the silylating agent in step c), a disiloxane is used in which all the residues R in the disiloxane are identical.

13. Process in accordance with at least one of the preceding claims, characterized by the feature that hexamethyldisiloxane is used as the silylating agent in step c).

14. Process in accordance with at least one of the preceding claims, characterized by the feature that the silylation process is carried out in a solvent.

15. Process in accordance with at least one of the preceding claims, characterized by the feature that the silylation process is carried out in the presence of a catalyst, preferably an acid.

16. Process in accordance with at least one of the preceding claims, characterized by the feature that the silylation process is carried out in the presence of catalytic quantities of trimethylchlorosilane.

17. Process in accordance with at least one of the preceding claims, characterized by the feature that, prior to step d), one washes the surface-silylated gel with a protic or aprotic solvent.

18. Process in accordance with at least one of the preceding claims, characterized by the feature that one subcritically dries the surface-silylated gel.

19. Process in accordance with at least one of the preceding claims, characterized by the feature that, prior to silylation, one reacts the gel obtained in step b) with a solution of an orthosilicate, which is capable of bringing about condensation, of formula $R^1_nSi(OR^2)_4$, preferably, an alkyl orthosilicate and/or an aryl orthosilicate, whereby $n = 2$ through 4 and R^1 and R^2 , independently of one another, are hydrogen atoms, linear or

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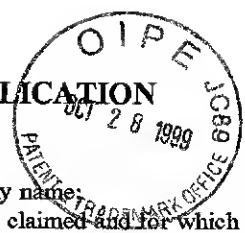
branched C₁-C₆ alkyl residues, cyclohexyl residues or phenyl residues, or with an aqueous silicic acid solution.

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SOLE DECLARATION FOR PATENT APPLICATION



As the below named inventor, I hereby declare that:

My residence, post office address and citizenship are as stated below next to my name;

I believe I am the original, first and sole inventor of the subject matter which is claimed and for which a patent is sought on the invention entitled METHOD FOR PRODUCING ORGANICALLY MODIFIED PERMANENTLY HYDROPHOBIC AEROGELS, the specification of which

☐ is attached hereto.

☒ was filed on May 25, 1999 as Application Serial Number PCT/EP97/06596.

I hereby state that I have reviewed and understand the contents of the above identified specification, including the claims, as amended by any amendment referred to above.

I acknowledge the duty to disclose information which is material to the examination of this application in accordance with Title 37, Code of Federal Regulations, § 1.56(a).

Prior Foreign Application(s)

I hereby claim foreign priority benefits under Title 35, United States Code, § 119 of any foreign application(s) for patent or inventor's certificate listed below and have also identified below any foreign application(s) for patent or inventor's certificate having a filing date before that of the application on which priority is claimed:

Country	Application Number	Date of Filing (day, month, year)	Date of Issue (day, month, year)	Priority Claimed Under 35 U.S.C. §119
PCT	PCT/EP97/06596	26 November 1997		

Prior United States Application(s)

I hereby claim the benefit under Title 35, United States Code, § 120 of any United States application(s) listed below and, insofar as the subject matter of each of the claims of this application is not disclosed in the prior United States application in the manner provided by the first paragraph of Title 35, United States Code, § 112, I acknowledge the duty to disclose material information as defined in Title 37, Code of Federal Regulations, § 1.56(a) which occurred between the filing date of the prior application and the national or PCT international filing date of this application:

Application Serial Number	Date of Filing (Day, Month, Year)	Status — Patented, Pending, Abandoned

And I hereby appoint, both jointly and severally, as my attorneys with full power of substitution and revocation, to prosecute this application and to transact all business in the Patent and Trademark Office connected herewith the following attorneys who are all members of the Bar of the District of Columbia, their registration numbers being listed after their names:

Donald W. Banner, Registration No. 17,037; Harold J. Birch, Registration No. 16,527; Edward P. McKie, Jr., Registration No. 17,335; William W. Beckett, Registration No. 18,262; Dale H. Hoscheit, Registration No. 19,090; Joseph M. Potenza, Registration No. 28,175; James A. Niegowski, Registration No. 28,331; Joseph M. Skerpon, Registration No. 29,864; Thomas L. Peterson, Registration No. 30,969; Nina L. Medlock, Registration No. 29,673; William J. Pisher, Registration No. 32,133; Thomas H. Jackson, Registration No. 29,808 and Peter D. McDermott, Registration No. 29,411

All correspondence and telephone communications should be addressed to Banner & Witcoff, Ltd. Eleventh Floor, 1001 G Street, N.W., Washington, D.C. 20001-4597, telephone number (202) 508-9100, which is also the address and telephone number of each of the above listed attorneys.

I hereby declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issuing thereon.

Signature Fritz Schwertfeger Date 25.8.99
Full Name of Sole Inventor SCHWERTFEGER, Fritz
Family Name First Given Name Second Given Name
Residence Germany
Citizenship German
Post Office
Address Raenenthaler Weg 32, D-60529 Frankfurt am Main, Germany DEX

BANNER & WITCOFF, LTD.
1001 G STREET, N.W.
WASHINGTON, D.C. 20001-4597
(202) 508-9100